Netot

Density Worksheet

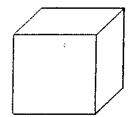
Name: Section:

Date:



1. Write the Density formula below.

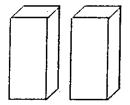
2. Calculate the density of the metal cube listed below. The mass is 8 grams and the volume is 4 ml. What is the density? Write the problem and answer below.



D =

3. Imagine that the cube is split in half exactly. (Use the numbers for MASS + Volume at

What is the mass of one half of the cube?

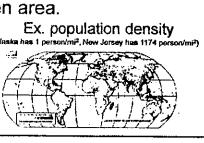


)=

4. Does size affect density?

- 5. Circle the less dense item.
 - a. wood or water
 - b. steel or water
 - c. helium or air
 - d. oil or water

Density is... the amount of in a given area. Ex. population density



Density

For science... density is equal to the mass of a thing divided by its volume.

D = ____ Mass – amount of _____ – measured in

Volume – amount of ___ something occupies - measured in or

Mass = D(v) v = m/d

Volume

For regularly shaped objects-_~ (cm)x (cm)x (cm)=cm;

For irregularly shaped objects- use (amount of water pushed aside equals the volume of the object)



Archimedes Principle

- Archimedes- Greek philosopher
- Was in his bath tub one morning he noticed that as his body went in to the water - the water rose out of the way. He realized that the volume of H2O that rose was equal to the
- So... the volume of an object is equal to the volume of _____ it displaces
- 1g H_2O = 1mL H_2O = 1 cm³ H_2O

Buoyancy



force that keeps things afloat. -in water: an object ____if its buoyancy is greater than its weight; __if its weight is greater than its buoyancy

-the buoyant force on the solid object is equal to the weight of the water displaced

Practice time

- If an object has a density <1g/cm3, what will it do if placed in water?
- If an object has a density >1g/cm3, what will it do if placed in water?
- If an object has a density =1g/cm3. what will it do if placed in water?