Name .				
	<u> </u>	<u></u>	Class	Date

ACTIVITY Atoms and Bonding

Bond Identification

- 1. When atoms share electrons to fill their outermost energy levels, they form bonds.
- 2. When atoms transfer electrons to fill their outermost energy levels, they form ______ bonds.
- 3. Indicate whether the atoms listed below will share electrons or transfer electrons. (*Hint:* When electrons are transferred, both atoms will have complete outermost energy levels. Otherwise, atoms share electrons.)
 - a. $:\ddot{O}\cdot + \cdot \ddot{O}:$
 - b. $K \cdot + \cdot \ddot{C}l$:

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- c. $\cdot\dot{C}\cdot + \cdot\ddot{O}$:
- d. $H \cdot + \cdot \ddot{F}$:
- e. $: \ddot{\mathbf{I}} \cdot + \cdot \ddot{\mathbf{I}}:$
- f. $:\ddot{S} \cdot + \dot{O}: + \dot{O}:$
- g. Na· + Na· + : \dot{S} ·
- h. Li \cdot + \cdot Cl:
- i. $Ca \cdot + \cdot Br : + \cdot Br :$
- j. $\cdot \dot{N}$: + $\cdot \dot{N}$:

4. Write the chemical formula and the electron-dot model for each of the products in question 3. If ions are formed, show the electric charge on each.

Chemical Formula	Election-Dot Model
a	
b	
c	
d	· ·
e	
f	
g	<u> </u>
h	
i	
•	