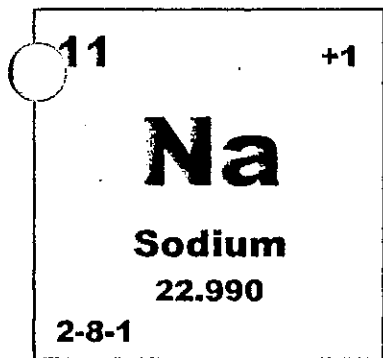


Name: _____ Date: _____ Period: _____



Bohr Model:

Atomic # _____

Protons (p⁺) _____

Neutrons (n⁰) _____

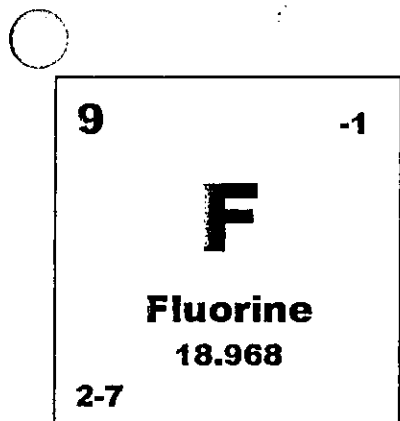
Electrons (e⁻) _____

Atomic mass _____

Valence (outer electrons) _____

Gains or loses e⁻? _____

Lewis Dot Structure:



Bohr Model:

Atomic # _____

Protons (p⁺) _____

Neutrons (n⁰) _____

Electrons (e⁻) _____

Atomic mass _____

Valence (outer electrons) _____

Gains or loses e⁻? _____

Lewis Dot Structure:

Name: _____ Date: _____ Period: _____

12	+2
Mg	
Magnesium	
24.305	
2-8-2	

Bohr Model:

Atomic # _____

Protons (p⁺) _____

Neutrons (n⁰) _____

Electrons (e⁻) _____

Atomic mass _____

Valence (outer electrons) _____

Gains or loses e⁻? _____

Lewis Dot Structure:

8	-2
O	
Oxygen	
15.999	
2-6	

Bohr Model:

Atomic # _____

Protons (p⁺) _____

Neutrons (n⁰) _____

Electrons (e⁻) _____

Atomic mass _____

Valence (outer electrons) _____

Gains or loses e⁻? _____

Lewis Dot Structure: