Derived Units

Directions Write the correct term on the line. 1. Density is found by dividing _____ by volume. 2. When you multiply or _____ measurements, you make a derived unit. 3. Density, area, and ______ are all derived units. 4. The SI unit for area is _____. Directions Write the letter of the answer to each question on the line. 5. If you cut an object in half, what happens to its density? C It stays the same. A It doubles. **B** It decreases by half. **D** It triples. 6. If a substance has a volume of 10 L and a mass of 5 kg, what is its density? **D** 5 kg/L **C** 0.5 kg/L **B** .50 kg/L A 2 kg/L 7. If density is mass divided by volume, what is mass? **D** $M = \frac{V}{D}$ A $M = \frac{\nu}{V}$ C M = D - V**B** $M=D\times V$ 8. Which of the following measurements has a derived unit? **A** $7.0 \, \text{m}^2$ **B** 7.0 kL **C** 0.97 dm **D** $0.97g/cm^3$ 9. If you are given mass in grams and volume in milliters, what will the unit of density be? \mathbf{B} cm³ $\mathbf{A} \mathbf{g} \times \mathbf{mL}$ C g/L D g/mL **Directions** Write the answer to each problem. Use the correct number of significant figures. 10. If an object has a mass of 23.1 g and a volume of 150 mL, what is its density? 11. What is the density of an object with a mass of 230 kg and a volume of 3,004 L? 12. What is the mass of an object with a density of 3.4 g/mL and a volume of 500.0 mL? 13. If an object has a mass of 3.0 g and a density of 0.023 g/mL, what is its volume? **14.** If a substance has a density of 0.23 g/mL, what is the mass of a 24-L sample? 15. A 10.0-g object with a density of 1.5 g/mL is cut in half. What is the volume

of half of the object?